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## Chemistry Chapter 7 Study Guide For Content Mastery Answers

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER  
Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III)

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sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

## 7 Chemical Formulas and Chemical Compounds

This Study Guide for Content Mastery for Chemistry: Matter and Change will help you learn more easily from your textbook. Each textbook chapter has six study guide pages of questions and exercises for you to complete as you read the text. The study guide pages are divided into sections that match those in your text.

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Chapter 7 Energy and Chemical Reactions. Review Skills 7.1 Energy Kinetic Energy Potential Energy Units of Energy Kinetic Energy and Heat Radiant Energy 7.2 Chemical Changes and Energy 7.3 Ozone: Pollutant and Protector Removal of UV Radiation by Oxygen and Ozone Molecules The Natural Destruction of Ozone 7.4 Chlorofluorocarbons: A Chemical Success Story Gone Wrong CFCs and the Ozone Layer.

## Chapter 7 Energy and Chemical Reactions

Chemistry Chapter 7 Study Guide. the force that holds two atoms together; may form by the attraction of a positive ion for a negative ion or by sharing electrons. a three-dimensional geometric arrangement of particles in which each positive ion is surrounded by negative ions and each negative ion is surrounded by positive ions; vary in shape due to sizes and relative numbers of the ions bonded.

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Chemistry - Chapter 7 Study Guide Section 7. 1: Valence electrons electrons in the outer energy level o responsible for bonding with other atoms How do you know how many valence electrons an element has? o Look at the group number Group 1A = 1 valence electron Group 7A = 7 valence electrons Etc...

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